

## Photocatalytic degradation of methyl orange dye by $\text{Ti}_3\text{C}_2\text{-TiO}_2$ heterojunction under solar light



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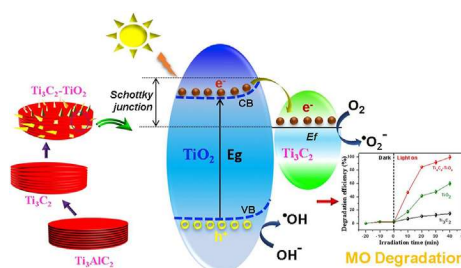
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### HIGHLIGHTS

- $\text{Ti}_3\text{C}_2\text{-TiO}_2$  heterostructure is synthesized from  $\text{Ti}_3\text{C}_2$  by hydrothermal oxidation.
- *In-situ* transformation of  $\text{TiO}_2$  leads to obtain  $\text{Ti}_3\text{C}_2\text{-TiO}_2$  heterostructures.
- $\text{Ti}_3\text{C}_2\text{-TiO}_2$  photocatalyst exhibits a significant efficiency for MO degradation.
- The proposed MO photocatalytic mechanism toward  $\text{Ti}_3\text{C}_2\text{-TiO}_2$  composite is illustrated.

### GRAPHICAL ABSTRACT



### ARTICLE INFO

#### Article history:

Received 28 December 2020

Received in revised form

22 February 2021

Accepted 1 March 2021

Available online 5 March 2021

Handling Editor: Derek Muir

#### Keywords:

$\text{Ti}_3\text{C}_2\text{-TiO}_2$  heterojunction

### ABSTRACT

Photocatalytic activity is a feasible solution to tackle environmental pollution caused by industrial pollutants. In this research,  $\text{Ti}_3\text{C}_2\text{-TiO}_2$  composite with a unique structure was fabricated successfully via a hydrothermal method. Especially, the *in-situ* transformation of  $\text{TiO}_2$  from  $\text{Ti}_3\text{C}_2$  MXene creates an intimate heterostructure, which leads to prolonging separation and migration of charged carriers. Thus, this  $\text{Ti}_3\text{C}_2\text{-TiO}_2$  composite enhances effectively methyl orange (MO) degradation efficiency (around 99%) after 40 light-exposed minutes. Besides, the optimal concentration of MO solution was estimated at 40 mg/L and  $\text{Ti}_3\text{C}_2\text{-TiO}_2$  photocatalyst also exhibited good stability after five runs. Moreover, the radical trapping test and the MO photodegradation mechanism over  $\text{Ti}_3\text{C}_2\text{-TiO}_2$  system were also demonstrated.

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